Synthesis and Magnetic Properties of Multiferroic $[C(NH₂)₃]Cr(HCOO)₃$ Metal-Organic Framework: The Role of Spin-Orbit Coupling and Jahn-Teller Distortions

Kunihiro Yananose,†,‡,[⊥] Ewan R. Clark,¶,[⊥] Paul J. Saines,[∗],¶ Paolo Barone,§ Alessandro Stroppa,[∗],[∥] and Jaejun Yu[∗],‡

†Korea Institute for Advanced Study, Seoul 02455, Republic of Korea ‡Center for Theoretical Physics, Department of Physics and Astronomy, Seoul National University, Seoul 08826, Republic of Korea ¶School of Chemistry and Forensic Science, University of Kent, Canterbury CT2 7NH, United Kingdom §Consiglio Nazionale delle Ricerche, Institute for Superconducting and Innovative Materials and Devices (CNR-SPIN), Area della Ricerca di Tor Vergata, Via del Fosso del Cavaliere 100, 00133 Rome, Italy ∥Consiglio Nazionale delle Ricerche, Institute for Superconducting and Innovative Materials and Devices (CNR-SPIN), c/o Department of Physical and Chemical Sciences,

University of L'Aquila, Via Vetoio I-67100 Coppito, L'Aquila, Italy \bot Contributed equally to this work

Abstract

We report for the first time the synthesis of $[C(NH₂)₃]Cr(HCOO)₃ stabilizing Cr²⁺ in a for$ mate perovskite, which adopts a polar structure and orders magnetically below 8 K. We discuss in detail the magnetic properties and their coupling to the crystal structure based on first principles calculations, symmetry, and model Hamiltonian analysis. We establish a general model for the orbital magnetic moment of the $[C(NH₂)₃]M(HCOO)₃$ (M = Cr, Cu) based on the perturbation theory, revealing the key role of the Jahn-Teller distortions. We also analyzed their spin and orbital textures in k -space, which show unique characteristics.

Introduction

Metal-organic frameworks (MOFs) are materials in which the metal ions are connected with each other by organic molecules. The choice of organic linkers allows a great variety in their crystal structures. One of its classes, the porous MOFs hold a large portion of cavities in them. Their tunable porosity enables applications in gas storage, catalysis, etc. Thus they are widely studied. 1,2 On the other hand, dense MOFs hold much smaller cavities in comparison to porous MOFs: Their cations are closer together and can thus play a significant role in the emergence of their functional properties. $3-5$ The combination of organic-inorganic features can induce both magnetism and ferroelectricity simultaneously, i.e., multiferroicity. In some multiferroic materials, the ferroelectric and magnetic orders

are coupled, giving rise to the possibility of controlling the magnetic property by the electric field, and vice versa. $6-8$ Thus, both the magnetic and electric orders, their coupling, and the role and control of the structural deformation are important aspects for dense MOFs. 9–24

Among dense MOFs, the $[C(NH₂)₃]M(HCOO)₃$ $(M = Mn, Fe, Co, Ni, Cu, and Zn)$ series have been synthesized with a perovskite-type ABX_3 structure.²⁵ They feature the guanidinium (Gua) ion $(C(NH_2)_3)^+$ on the A site, the 3d transition metal ions (M^{2+}) occupying the B site, and formate HCOO[−] ion as the X site anions as shown in Fig. 1 (a) and (b). These materials show magnetic ordering of their M^{2+} ions; notably, only the Cu-based analog (hereafter denoted as 1-Cu), featuring the Jahn-Teller (JT) active ion $Cu^{2+} (d^9)$, displays a polar structure and a weak ferromagnetic component that has been proposed to correlate with JT distortions.^{9,25} Similarly, a theoretical study on the not yet synthesized perovskite MOF with $M = Cr^{2+}$ (hereafter denoted as **1-Cr**) proposed a non-trivial role of the Cr^{2+} $(d⁴)$ JT active ion in shaping both ferroelectric and (weak) ferromagnetic properties. ¹² From one hand, the non-polar JT distortions couple to another non-polar mode giving rise to an inversion symmetry breaking hybrid mode, resulting in a so-called hybrid improper ferroelectricity. $26-29$ On the other hand, the weak ferromagnetic component was proposed to arise from the JT-related orbital ordering and its interplay with spin-orbit coupling (SOC) . ¹² As opposed to most of their inorganic counterparts, the multiferroic phase of ABX_3 1-Cu and 1-Cr holds the promise of a stronger coupling between magnetic and ferroelectric properties, as both functional properties share the JT activity as a common origin. 9,12,14 Although the 1-Cr was theoretically predicted to display multiferroic properties like the similar $1-Cu$, it was not previously possible to synthesize it due to the significant difficulties encountered in stabilizing Cr^{2+} , which undergoes rapid oxidation in air to the more stable trivalent oxidation state. Furthermore, recent experimental studies of multiferroic Cr^{2+} halides with layers³⁰ or chains³¹ of Cr octahedra further emphasize the

potential for unexploited functionality in Cr^{2+} hybrid perovskites.

In this study, we successfully synthesized 1- Cr for the first time, confirmed its polar structure, and measured its magnetic properties. The structural analysis confirms the theoretical predictions, while we observe an antiferromagnetic transition at $T_c \sim 8$ K, roughly twice the Néel temperature of 1-Cu but four times smaller than the theoretical prediction, with no evident signatures of weak ferromagnetism (WFM). We accordingly revised the earlier magnetic model¹² by proposing a new and more accurate set of magnetic parameters extracted from first-principles calculations, which are able to reproduce the experimental measurements more accurately as well as the magnetic transition temperature. Noticeably, the revised model which is based on the subtle interplay of orbital order, SOC, and structural distortions is able to capture the complex magnetic configuration of 1-Cr. The key role of SOC, so commonly neglected in 3d transition metals, in the magnetic properties of 1-Cr and 1-Cu, is emphasized by the significant contribution this makes to their effective magnetic moments; particularly in **1-Cu**, where the orbital ferromagnetic component is even comparable to the spin contribution. We explain the orbital magnetic moment, previously neglected in the $1-Cr/Cu$, and its dependence on orbital ordering based on the second order perturbation theory and the JT effective Hamiltonian. Our microscopic model is in very good agreement with the density functional theory (DFT) calculation results and could be applied to similar compounds. Finally, we also analyze the spin and orbital textures in k -space, which show unique characteristics.

Methods

All synthetic procedures were performed according to standard Schlenk line procedures under an atmosphere of dry argon. Isolated Cr^{2+} samples were stored in an argon glovebox, and samples for powder X-ray diffraction and SQUID magnetometry analysis were made up in the same glovebox. This was essential to prevent the oxidation of Cr^{2+} to the more stable trivalent oxidation state. Caution! Chromium salts are known sensitizers and so care must be taken to avoid generating any loose dust and to dispose of all samples appropriately. Multiple reactions described herein involve release of a condensable gas whilst working on a Schlenk line; care should be taken both to avoid sealing reaction vessels to avoid pressure build-up, and to ensure that cryogenic traps can vent safely on warming.

Cyan blocklike crystals of 1-Cr suitable for structural determination were made by layering a colorless solution of $\rm [C(NH_2)_3]/HCOO$ (0.420) g, 4 mmol) (See Sec. S1 of Supporting Information (SI) for preparation method) and H_2COO (165 μ l, 4 mmol) in 50:50 v/v water/ethanol upon a pale blue solution of $CrCl₂$ (61 mg, 0.5) mmol) in water (2 cm^3) . This produced an intense purple interface with crystals formed after standing for a week. A bulk sample suitable for further analysis was made by the addition of an aqueous solution of $CrSO₄·5H₂O$, made using literature methods, 32 (0.5 M, 5 cm³) into an aqueous solution of $\rm [C(NH_2)_3]/HCOO$ (See Sec. S1 for further details) in a single portion. This was stirred briefly to give a homogenous purple solution which, on standing overnight, transformed to a pale blue supernatant above a cyan solid. The solid was isolated by filtration, washed with EtOH $(2 \times 10 \text{ cm}^3)$, and dried in vacuo to give $1-Cr$ as a cyan microcrystalline powder (0.541 mg, 2.1 mmol, 84 % yield.)

Structure determination was carried out using single crystal X-ray diffraction (SCXRD) data recorded on an Agilent SuperNova Dual diffractometer (See Sec. S2 for further details of experimental and structure solution methods). Structures from this experiment are deposited in the CSD as entries 2278325-2278331. Bulk purity of the sample was assessed using powder X-ray diffraction (PXRD) collected using a Rigaku Miniflex using Cu K α (40 kV, 15 mA) with the sample mounted in a propriety air-sensitive sample holder. A Le Bail fit was carried out using the program Rietica.³³ Elemental microanalysis was carried out at London Metropolitan University. Magneto-

metric studies were performed using a Quantum Design MPMS 7 magnetometer, utilizing an applied field of 1000 Oe for variable temperature susceptibility (χ) measurements. Samples were placed in gelatin capsules enclosed inside a pierced straw with a uniform diamagnetic background. Differential Scanning Calorimetry (DSC) measurements were performed on 1-Cr using a NETZSCH DSC 200PC with the sample in a closed Al pan. Sample loading and data collection were performed under an inert nitrogen atmosphere.

We used the Vienna Ab-initio Simulation Package $(VASP)^{34-37}$ for the first-principles DFT calculation. To include SOC, we performed non-collinear spin DFT calculations. Generalized gradient approximation by Perdew-Burke-Ernzerhof (GGA-PBE) for the exchange-correlation functional³⁸ and the projector augmented wave pseudo-potential ³⁹ were adopted. The plane wave energy cut-off was chosen to be 500 eV. $4 \times 4 \times 4$ regular k-space grid was used. For the lattice constants, experimental values of 1 -Cu $a = 8.5212$ Å, $b = 9.0321$ Å, and $c = 11.3497$ Å from Ref. 25 were used for both 1-Cr and 1-Cu for consistency with the previous theoretical works. 9,12 Note that the measured lattice constants of the 1-Cr are close to these values, thus justifying our initial guess (See Sec. S2). For a ferroelectric structure, one can define a corresponding paraelectric virtual structure of higher symmetry, referred to as pseudo symmetric structure, using the group-theoretical method implemented in the software PSEUDO of the Bilbao Crystallography server. ⁴⁰

We estimated the transition temperature (T_c) by using the spin model suggested in Ref.12 and the Monte Carlo simulations adopting a standard Metropolis algorithm. We additionally considered the on-site Coulomb energy correction in DFT (DFT+ $U+J$ calculation⁴¹), which was neglected in previous calculations, ¹² when obtaining the spin model parameters. Two sets of parameters, $(U, J) = (2.5, 0.5)$ and $(3.0, 1.0)$ (eV) were used. Further details of the spin model can be found in Sec. S4 and S5 of SI.

Figure 1: (a,b) Crystal structure of 1-Cr at 300 K seen from different directions indicated by crystal axes in each panel with the Cr octahedra shown in blue and carbon, hydrogen, nitrogen, and oxygen atoms shown as black, pink, light blue, and red spheres, respectively. (c) Le Bail fit to a powder X-ray diffraction pattern of 1-Cr with the experimental data shown as black crosses, the calculated and difference intensities shown as red and green lines, respectively, and the vertical blue markers noting the expected position of Bragg reflections. The omitted peak from the fit has been confirmed as being associated with the air-sensitive sample holder, likely from a Ni-based alloy. R_p , $R_{\rm wp}$ and χ^2 of the fit were 5.89 %, 9.81 % and 8.29.

Results and Discussion

Synthesis and Crystal Structure

Initial syntheses of 1-Cr followed a modified preparation from those used for the Co and Fe analogues, ²⁵ but yields were low, and sample purity was poor as a result of contamination from Cr^{3+} within the commercial source. Nevertheless, slow diffusion of a solution of $|C(NH_2)_3|$ (HCOO) in ethanol into a solution of chromous/chromic chloride in water allowed the growth of single crystals of sufficient quality for SCXRD studies. Attempts to scale this method up produced inseparable mixtures of crystals, attributed to the Cr^{3+} impurities in the starting material, as suggested by the pale color of the Cr solution since fresh $Cr^{2+}Cl_2$ solutions are an intense royal blue. To avoid this problem, efforts to make bulk samples of 1-Cr for further analysis utilized $CrSO_4.5H_2O$ in an aqueous medium. Fits to PXRD patterns indicated this method resulted in a sample with a structure consistent with that determined from SCXRD with only trace amounts of an unidentified impurity phase (See Fig. 1 (c)). Lattice parameters were determined to be $a = 8.6358(8)$ Å, $b = 11.6480(15)$ Å and $c = 9.1050(9)$ Å yielding a unit cell volume of $915.87(15)$ \AA^3 . The purity of this sample was further confirmed by elemental analysis results (experimental values C 19.47 %, H 3.57 % and 17.54 % N to calculated values of 19.44 %, 3.67 % and 17.00 %, respectively).

Examination of the systematic absences of SCXRD data of 1-Cr collected at 100 K indicates it adopts a structure in either $Pna2₁$ or *Pnma* orthorhombic symmetry, with the systematic absences required for the P *nan* structure adopted by the $[C(NH_2)_3]M(HCOO)_3$ (M $=$ Mn-Ni and Zn) compounds violated. ²⁵ Extensive attempts to solve the structure were only successful for $P na2₁$ symmetry with $P nma$ not giving a chemically sensible or even complete solution. This is consistent with the previous report of **1-Cu** adopting $P na2₁$ symmetry and previous combined first-principles calculations and group theoretical analysis of 1- Cr indicating this to be the expected symmetry of this material. 9,12,25 We should note the crystal used in this study was a twin by in-

Figure 2: Differential scanning calorimetry trace of **1-Cr** measured up to 600 \degree C. The insert highlights the signal at lower temperature.

version, which may complicate using them to experimentally confirm ferroelectric switching.

As expected, 1-Cr adopts a hybrid perovskite structure which closely resembles other $[C(NH₂)₃]M(HCOO)₃$ phases, including a similar conventional $a^-a^-c^-$ tilt system²⁵ (See Fig. 1 for crystal structure and Fig. S1 in SI for asymmetric unit). As previously seen in isostructural **1-Cu** where the d^9 Cu²⁺ cations also have a JT distorted bonding environment (See Table S2 for bond angles), the elongated axis of the d^4 Cr²⁺O₆ octahedra also alternates within the ab-plane. This is in a Cr- $O_{short}...Cr$ - $O_{long}...Cr$ - $O_{short}...Cr$ - O_{long} pattern between neighboring octahedra connected by formates along both orthogonal directions in this plane; at 300 K, the Cr-Oshort distances in this plane are $2.031(9)$ and $2.066(8)$ Å while the Cr- O_{long} distances are 2.347(9) and 2.390(8) Å (See Fig. S2 for evolution with temperature). The remaining two $Cr-O_{short}$ bond distances required to complete a conventional trans-elongated JT distorted octahedra are oriented along the c-axis with distances of 2.065(14) and 2.079(15) Å. The bond valence sum of the Cr cation is 2.00 consistent

with the expected divalent oxidation state.⁴² Along with the JT distortion of the octahedra in 1-Cr, the main difference in the structure of this phase when compared to the P nan $[C(NH₂)₃]M(HCOO)₃$ frameworks is the subtle rotations of the Gua cations along the a and c -axis.^{12,25}

As is observed for other $[C(NH₂)₃]M(HCOO)₃$ phases, variable temperature SCXRD analysis indicates the Gua remains ordered up to 400 K, the highest temperature measured during this study. ²⁵ At 400 K, a significant decrease in data quality of variable temperature SCXRD is noted (average I/σ declines from being consistently >15.9 to 11.9 at 400 K), which may indicate that the measurements are approaching a temperature at which the material begins to lose crystallinity. At this temperature, the reflections indicating the violation of the systematic absences associated with the second n -glide required for *Pnan* symmetry are lost. While this is most likely a result of the poorer data quality available from this air-sensitive sample above ambient temperature there is a significant decrease in the JT distortion at the same temperature (See Fig. S2), which could alternatively indicate the onset of a transition to P nan symmetry.

DSC analysis of **1-Cr** indicates a possible phase transition near 140 ◦C with an enthalpy change of approximately -0.9 kJ mol^{-1} . This may suggest the weakening in the JT distortion of 1-Cr identified in the crystal structure at 400 K may be a precursor to a phase transition related to the JT distortion. The enthalpy change measured here is at least an order of magnitude lower than the JT energies of $[Cr(H₂O)₆]²⁺$ suggested by calculations and spectroscopic measurements. 43,44 This suggests that if this transition is related to an apparent loss of the JT distortion in the crystal structure it is likely related to a loss of orbital order rather than quenching of the JT effect, i.e., a loss of ordering between the JT distorted local structures. There are three further peaks indicative of endothermic processes in 1-Cr at higher temperatures, centred near 249 ◦C, 272 ◦C and 335 ◦C with estimated enthalpy changes of -7.0 kJ mol⁻¹, -65.2 kJ mol⁻¹ and -4.3 kJ mol⁻¹.

Figure 3: (a) Plot of χ T versus temperature for **1-Cr** in a 1000 Oe field. The insert shows ZFC and FC χ measurements measured between 5 and 20 K and the predicted χ by the MC simulation. (b) Plot of $1/\chi$ versus temperature for **1-Cr**, which is well fitted by a linear trend line (shown in blue) from 20-300 K. (c) Plot of magnetization, M, of 1 -Cr versus applied field, H, at 5 K.

The temperatures at which these features appear are broadly consistent with the reported decomposition temperatures of the Mn, Fe, and Co analogs so we anticipate these features may also indicate the decomposition of $1-Cr$.²⁵

The thermal expansion of this material across the 100-400 K range studied in this study is anisotropic with significant positive thermal expansion of 50(5) and 29(7) MK⁻¹ observed along the a -axis and c -axis while a modest negative thermal expansion of $-14(3)$ MK⁻¹ is measured along the b-axis (See Fig. S3 for lattice parameter plot). This is consistent with the orientation of analogous anisotropic thermal expansion observed from other members of the $\rm [C(NH_2)_3]M(HCOO)_3$ series, ⁴⁵ appearing to be of a similar scale to the Mn analog which has the highest anisotropic negative thermal expansion of the members reported experimentally thus far, and related $[C(NH₂)₃]$ Er(HCOO)₂(C₂O₄).⁴⁶ Framework hinging combines with the expansion of the M-formate-M struts upon heating to govern the overall observed thermal expansivities. For other members of the $[C(NH_2)_3]M(HCOO)_3$ series, the anisotropic thermal expansion has been attributed to hinging of the metal-formate framework as the pore shape becomes more isotropic on heating. $45,46$ In the case of 1-Cr, this leads to greater expansion of the a -axis, along which the hinging angles are acute, accompanied by a more modest contraction along the b-axis, along which the hinging angles are obtuse (See Fig. S4). The connectivity of the octahedra along the c-axis by the formate ligands means that expansion in this direction is only controlled by the strut expansion as the framework cannot hinge in this direction, unlike the ab-plane where the connectivity via the formate ligands is approximately along the <110> directions.

Magnetic Properties

Field cooled (FC) and zero field cooled (ZFC) χ measurements with respect to temperature in a 1000 Oe applied magnetic field indicate maxima just below 8 K, with no divergence of these measurements below this temperature (See Fig. 3 (a)). This indicates a transition to a compensated antiferromagnetic (AFM) state. A plot of $1/\chi$ with temperature is well fitted by a linear trend between 20 K and 300 K with deviations observed below 20 K consistent with AFM correlations (See Fig. 3 (b)). This fit yields a Weiss constant, Θ , of -9.8 K, consistent with the observed AFM transition slightly below this. This fit also indicated an effective magnetic moment of 4.70 μ_B , modestly below the spin-only magnetic moment expected for

a high spin d^4 cation, 4.90 μ_B , but consistent with the reduced value observed for other scarce examples of octahedral Cr^{2+} compounds with group 16 based ligands, including $Cr_3(PO_4)_2$ $(4.50 \ \mu_B)^{47}$ and Y₂CrS₄ (4.70 μ_B).⁴⁸ χ T gradually declines from a value consistent with noninteracting Cr^{2+} cations at 300 K to 90 K before decreasing more rapidly to 5 K, consistent with strong AFM coupling emerging at low temperatures (See Fig. 3 (a)). Consistent with an AFM state, isothermal magnetization measurements at 5 K gradually increase as a function of the applied magnetic field, reaching a value of 1.10 μ_B per Cr^{2+} cation at 50 kOe, with no indication of hysteresis observed (See Fig. 3 (c)). Determining the magnetic structure of 1-Cr using neutron diffraction would be desirable future work in order to gain more insights into the magnetic structure at very low temperatures. This is not a trivial task due to the need to make more than a gram of a perdeuterated sample of this air-sensitive material.

First-Principles Calculations and Model Study of Magnetic Properties

Definition of structural interpolation

The space group symmetry of JT distorted 1- Cr/Cu is the polar $Pna2₁$ (No. 33) with corresponding centrosymmetric pseudo-symmetric group P nan (No. 52, P nna in standard settings). The structural distortion relating the *Pnan* to $Pna2₁$ structure can be expressed using the interpolation parameter $\lambda^{9,12}$ ($\lambda = 0$ for the *Pnan* and $\lambda = 1$ for the original *Pna* $2₁$ structure). We denote the atomic positions at $\lambda = 0$ as \mathbf{r}_{Pnan} , and the displacement vectors from $\lambda = 0$ structure to $\lambda = 1$ structure as **u**. Then the atomic positions of the interpolated structure are written as $\mathbf{r}(\lambda) = \mathbf{r}_{Pnan} + \lambda \mathbf{u}$. The structural path defined by λ evolving from +1 to −1 represents a polarization switching path, passing through a centrosymmetric P nan $(\lambda=0)$. Moreover, the net magnetic moment also reverses its direction along the path^{9,12} as reproduced in Fig. 4. Notably, for the 1-Cr, the predicted structure¹² is in good agreement

Figure 4: Spin, orbital, and total (S+L) magnetic moment per unit cell of (a) 1-Cu and (b) **1-Cr** with respect to the structure parameter λ . For the **1-Cu(1-Cr)**, $a(c)$ -component is shown.

with the new experimental structure in terms of space-group symmetry, JT distortions, and structural details (See Sec. S2).

Magnetic properties

As the unit cell of $P na2_1$ **1-Cr/Cu** comprises four magnetic ions, we can define four magnetic order parameters per cell compatible with nonmagnetic translational symmetries (See Sec. S3), describing a ferromagnetic configuration M and three antiferromagnetic configurations G, A , and C . While all magnetic moments are antiparallel in the G-type AFM configuration, the A-type AFM (A-AFM) structure is characterized by layers of parallel magnetic moments antiferromagnetically aligned, and the C-type comprises ferromagnetic chains that are antiferromagnetically coupled (See Fig. S5).

In the $1-Cr/Cu$, antiferro-distortive ordering of the octahedra induces an orbital ordering, i.e., the cooperative JT effect determines the orbital structure characterized by intralayer antiferro-orbital ordering and interlayer ferro-orbital ordering. Accordingly, the Goodenough-Kanamori-Anderson rule 49,50 predicts the FM interaction between the in-plane neighboring ions and AFM interaction between the out-of-plane neighboring ions, i.e., A-AFM. Our new $DFT(+U+J)$ calculations indeed confirm this scenario, in agreement with previous studies. 9,12 Noticeably, a weak ferromagnetic component, appearing as a secondary order parameter, is allowed only for $P n^{\prime} a^{\prime} 2_1$:

 ${A_a; G_b, M_c}$ and $Pna'2'_1$: ${A_c; C_b, M_a}$, where the prime means that the symmetry operation is accompanied by the time-reversal operation. The spin axes of AFM alignment identified by the new calculations are crystallographic c-axis for **1-Cu** $(Pna'2'_1)$ and a-axis for 1-Cr $(Pn'a'2_1)$, consistently with previous studies. $9,12$ However, there exist also secondary AFM orders G_b and C_b , which were previously overlooked. 9,12 A weak ferromagnetic moment is then allowed by the magnetic space symmetries to develop along the a -axis in **1-Cu** and along the c -axis in 1 -Cr, which can be viewed as resulting from a small canting of the primary A-AFM configuration. The experiments observed the WFM moment of the $1-Cu, ^{25,51}$ while no evident signature for WFM appears from our new measurements for the **1-Cr**. This discrepancy motivated us to revisit the magnetic model proposed earlier, by using more accurate parameters as derived from the present work.

The spin canting can be attributed to the magnetic single-ion anisotropy (MSIA) originating from $SOC¹²$ ruling out the antisymmetric Dzyaloshinskii-Moriya exchange interaction since it is incompatible with the $\{A_a, M_c\}$ and ${A_c, M_a}$ coupling (See Sec. S4 and S10). The antiferro-orbital ordering within the ab-plane is responsible for different anisotropy axes on neighboring magnetic sites, which may cause a canting of the primary magnetic configuration.^{12,52} The rotation of the orbital order/JT distortion pattern, accompanying the switching of λ from 1 to -1, which was shown in Ref. 12, can then be naturally related to a rotation of the local anisotropy axes and a corresponding switching of the magnetic moment, as shown by the DFT calculations (Fig. 4).

The calculated WFM spin magnetic moments are shown in Fig. 4. The total spin moment in the 1-Cr (0.04 μ_B per each magnetic atom, with a canting angle of about 0.63°) is significantly smaller than the previous estimate $(1 \mu_B)$, 14.5°).¹² The present calculations emphasizing a very small canting angle are more consistent with that the experiment in this work could not see the WFM $(Fig. 3(c))$.

Here, we adopt the same classical spin model introduced in Ref. 12. The model comprises

Figure 5: Monte Carlo calculations. Susceptibility associated to primary order parameter A-AFM (a) and specific heat (b) as a function of temperature and calculated from two sets of model parameters extracted from $DFT+U+J$ computations. Both quantities display sharp peaks signaling transition temperatures in a range comprised between 9.5 and 14.5 K. Panel (c) displays the evolution of order parameters A_a, G_b , and M_c below the transition temperature for the set of parameters corresponding to $(U, J) = (3, 1)$ eV.

the competition between a standard Heisenberg term $H=\frac{1}{2}$ $\frac{1}{2} \sum_{ij} J_{ij} S_i S_j$, with an intralayer ferromagnetic exchange $J_{ab} < 0$ and interlayer antiferromagnetic $J_c > 0$, and a site-dependent MSIA term. The latter can be written in a local reference frame defined by M-X bonds as $H_{\rm sia} = E \sum_i \left[(\boldsymbol{S}_i \cdot \boldsymbol{e}_i^s)^2 - (\boldsymbol{S}_i \cdot \boldsymbol{e}_i^l)^2 \right] + D \sum_i (\boldsymbol{S}_i \cdot \boldsymbol{e}_i^l)^2$ $(e_i^m)^2$, with \overline{E} , D denoting the principal values of the MSIA tensor and e_i^l, e_i^m, e_i^s denoting the long, medium and short M-X bonds of the JT-distorted MX_6 octahedra. We estimated the model parameters from the totalenergy mapping of DFT computations with different collinear and canted magnetic configurations (See Sec. S4 for details). When comparing with previous estimates 12 obtained within a bare GGA approach, we observe a substantial reduction of both J_{ab} and J_c exchanges (as expected by the inclusion of the U correction) and, most prominently, of the MSIA parameter E (See Table 1).

In addition, we performed Monte Carlo simulations using the new parameters (See Sec. S5

Table 1: Parameters of the spin model

Model parameters (all couplings in meV, tilting angle θ_t in degrees) used in the Monte Carlo simulations for the 1-Cr, with and without the $+U+J$ corrections. The parameters in the previous work¹² are also shown for comparison. Notice that the estimate for the angle θ_t has been obtained from magnetic anisotropy energies, 12 the fact that it is the same in the two sets is a non-trivial result: such estimate is in good agreement also with the structural tilting angle (given in brackets). The values in brackets for the parameter E are the estimates obtained by using the structural tilting angle θ_t , showing small deviations that are found not to affect the results significantly.

	(U,J)	J_{α}	J_{ab}	IJ	F	θ_{t}
no U (Ref. 12)	(0, 0)	0.824	-0.453 0.113		0.745	30.8
no U (this work)	(0,0)	0.837	-0.459	0.092	0.035(0.036)	33.6(31.9)
set 1	$(2.5, 0.5)$ 0.475 -0.111 0.115				0.038(0.041)	33.8(31.9)
set 2	$(3.0, 1.0)$ 0.453 -0.042 0.149				0.059(0.062)	33.8 (31.9)

for further details). Results are summarized in Fig. 5, where the evolution with temperature of specific heat, order parameters, and susceptibility of the primary A-AFM order parameter are shown. The magnetic transition is estimated between 9.5-14.5 K, in fairly good agreement with the experimental data (~ 8 K). The primary order parameter is the A-AFM, as shown by the sharp peak in the susceptibility, aligned along the a -axis. A WFM aligned along the c axis also arises under T_c as a secondary order parameter, consistently with the prediction by DFT , 12 even though it is strongly renormalized, while the other competing secondary parameter, corresponding to a G-AFM canting parallel to the b axis, is found to be substantial (See Sec. S4). The dominating antiferromagnetic interactions can also be deduced by the powderaverage magnetic susceptibility, shown in the inset of Fig. 3 (a) (See Fig. S9 for the full data) and again in good agreement with the experimental data.

In this study, we calculate and model the orbital magnetic moment in $1-Cr/Cu$, which was neglected in previous studies^{9,12} in more detail and is not generally accounted for in the studies of metal-organic formates. Here, we show that it may have a non-negligible contribution to the total magnetic moment. Generally, the magnetic moment has spin and orbital contributions. For a transition metal ion in an octahedral environment, the orbital magnetic moment is quenched when the t_{2g} d-orbitals are fully- or half-filled. However, the presence of

the SOC can still induce a small orbital magnetic moment. In the case that the spin contribution to the ferromagnetic component is much smaller than $1 \mu_B$ per magnetic ion, such as the case of WFM with a tiny canting angle, the orbital contribution to the ferromagnetic component may become significant to it. Indeed, we found that in 1-Cu, the orbital ferromagnetic component is collinear and comparable in magnitude to the spin magnetic moment, which the latter is not commonly expected for the orbitalquenched d^9 case. In the range of large $|\lambda|$, the orbital contribution (0.070 μ_B per unit cell at $\lambda = 1$) to the ferromagnetic moment is larger than the spin contribution (0.043 μ_B). On the other hand, for the 1-Cr, the contribution of the orbital moment to the ferromagnetic net moment $(-0.024 \mu_B)$ is much smaller in comparison with the contribution from the spin moment $(0.172 \,\mu_B)$ and they are anticollinear. Fig. 4 (a) and (b) show the calculated spin and orbital magnetic moment of 1 -Cu along a and 1 -Cr along c direction, respectively. Notably, the experimental effective magnetic moment for 1-Cr is lower than the spin-only magnetic moment while that of **1-Cu** is notably higher than its spin-only moment. ²⁵ These are reminiscent of Hund's third rule of the atomic limit: The spin and orbital moments are anticollinear for the less-than-half-filled case (1-Cr), but aligned for the more-than-half-filled case $(1-Cu)$. In the next section, we define a model of the orbital angular momentum in $1-Cr/Cu$ and its interplay with JT distortion/orbital ordering which

perfectly accounts for the observed trends.

Model for Orbital Magnetic Moment in 1-Cr/Cu

In order to explain the orbital magnetic moment in $1-Cr/Cu$, we introduce a model based on perturbation theory and JT effective Hamiltonian within a single ion description, which is an improvement of the previous work. ¹² The perturbation approach for the orbital angular momentum and MSIA corresponds to the Bruno theory, $53,54$ but we ignore the k-space dispersion for simplicity. The SOC Hamiltonian is written as $H_{\text{SOC}} = \zeta \mathbf{S} \cdot \mathbf{L}$, where **S** and **L** are spin and orbital angular momentum operator, respectively. We will consider only d-orbitals here.⁵⁵

In the O_6 octahedron cage, the crystal field splits d-orbitals into lower energy t_{2g} orbitals (d_{yz}, d_{zx}, d_{xy}) and higher energy e_g orbitals $(d_{x^2-y^2}, d_{z^2})$. In addition, the JT effect splits the degeneracy of e_q orbitals by deforming the $\overline{\mathrm{O}}_6$ cage.^{50,56} This deformation is represented by two distortion modes $Q_2 = (1/\sqrt{2})(l_x - l_y)$ and $Q_3 = (1/\sqrt{6})(2l_z - l_x - l_y)$, where the l_i means the distance from the center to the oxygen on the i-axis. The JT distorted structure is expressed with the JT phase θ_{JT} as $|\theta_{\text{JT}}\rangle =$ $\cos \theta_{\text{JT}} |Q_3\rangle + \sin \theta_{\text{JT}} |Q_2\rangle$ and $\tan \theta_{\text{JT}} = Q_2/Q_3$. The JT effective Hamiltonian taking the e_g orbitals as a basis is given by

$$
H_{\rm JT} = \gamma \begin{pmatrix} q_1 & q_2 \\ q_2 & -q_1 \end{pmatrix} + \frac{1}{2} C q^2 \mathbf{I}_2 \tag{1}
$$

where $q_1 = q \cos \theta_{\text{JT}}$, $q_2 = q \sin \theta_{\text{JT}}$, and \mathbf{I}_2 is $2 \times$ 2 identity matrix. $\overline{50,57-59}$ The energy eigenvalues are $E_{\pm} = \pm \gamma q + \frac{1}{2} C q^2$ and eigenstates are

$$
|d_{-}(\theta_{\text{JT}})\rangle = -\sin(\theta_{\text{JT}}/2)|d_{x^2-y^2}\rangle + \cos(\theta_{\text{JT}}/2)|d_{z^2}\rangle
$$

$$
|d_{+}(\theta_{\text{JT}})\rangle = \cos(\theta_{\text{JT}}/2)|d_{x^2-y^2}\rangle + \sin(\theta_{\text{JT}}/2)|d_{z^2}\rangle.
$$

(2)

These unitary rotated e_g -orbitals rewrite the orbital angular momentum operator and, consequently, the SOC Hamiltonian H_{SOC} (See Sec. S7 for details).

The perturbation theory is applied to ob-

tain the orbital angular momentum induced by SOC. The d-orbitals with the 'JT rotated' e_q orbitals (Eq. (2)) are taken as the unperturbed basis. Then, the orbital angular momenta are obtained as expectation values of the orbital angular momentum operators defined with the JT rotated e_q -orbitals up to first order in ζ (See Sec. S8 for the details). As a result, the orbital angular moments of the d^4 (1-Cr) configuration are

$$
\langle L_x \rangle_{d^4} = -\zeta \frac{\left(\cos\left(\frac{\theta_{\text{JT}}}{2}\right) + \sqrt{3}\sin\left(\frac{\theta_{\text{JT}}}{2}\right)\right)^2}{E_+^0 - E_{yz}^0} s_x
$$

$$
\langle L_y \rangle_{d^4} = -\zeta \frac{\left(\cos\left(\frac{\theta_{\text{JT}}}{2}\right) - \sqrt{3}\sin\left(\frac{\theta_{\text{JT}}}{2}\right)\right)^2}{E_+^0 - E_{zx}^0} s_y \quad (3)
$$

$$
\langle L_z \rangle_{d^4} = -\zeta \frac{\left(2\cos\left(\frac{\theta_{\text{JT}}}{2}\right)\right)^2}{E_+^0 - E_{xy}^0} s_z
$$

for each x, y, z component, where \hat{s} = (s_x, s_y, s_z) is the local spin direction with respect to the local orbital coordinates aligned to the octahedron. We adopt the atomic units in which $\hbar = 1$. For d^9 configuration (1-Cu), $\langle L_i \rangle_{d^9} = - \langle L_i \rangle_{d^4}$. It is noteworthy that the resultant local orbital moment is not always parallel to the spin. The deviation is explicitly determined by θ_{JT} .

The orbital angular momentum formula, Eq. (3), can be used for the orbital magnetic moment by simply replacing the angular momentum with the magnetic moment in Bohr magneton μ_B unit for both the spin and or-

Table 2: Symmetry operations in $Pna'2'_1$ and $Pn'a'2_1$

 $\langle x^2 \rangle$ ^{the} identity operation. Subscripts a, b, and c Symmetry operations and the transformation rules of the magnetic moment (L) in the magnetic space group $Pna'2'_{1}$ and $Pn'a'2_{1}$. 1 means are crystallographic axes.

bital as we use the atomic units. As stated in the previous section, two magnetic groups, $Pn'a'2₁$ and $Pna'2'₁$ are considered for both **1-Cr** (d^4) and **1-Cu** (d^9) . For simplicity, we will ignore the spin canting in the following analysis. Once the local magnetic moment at a reference Cr/Cu site is obtained by Eq. (3), the magnetic space group symmetry determines the magnetic moments at other sites according to the rules listed in Table 2. For the details, see Sec. S9 of SI.

For the d^4 configuration (Cr^{2+}) with $Pn'a'2_1$ symmetry, the total orbital moment is

$$
\langle L \rangle_{\text{total}}^{Pn'a'21} = -2 \left(\frac{\zeta}{E_+^0 - E_{yz}^0} - \frac{\zeta}{E_+^0 - E_{zx}^0} \right) \times \sin \theta_t (2 - \cos \theta_{\text{JT}}) - 2\sqrt{3} \left(\frac{\zeta}{E_+^0 - E_{yz}^0} + \frac{\zeta}{E_+^0 - E_{zx}^0} \right) \sin \theta_t \sin (4)
$$

along the c-axis, where θ_t is the tilting angle of the O_6 octahedron from the c-axis. Since the difference between t_{2g} orbitals E_{yz}^0 and E_{zx}^0 is smaller than the crystal field splitting, the first term is small compared to the second term. Moreover, $\langle L \rangle_{\text{total}}$ vanishes when $\theta_{\text{JT}} = \pi$, or $\lambda = 0$, since $l_x \approx l_y$ leads to $E_{yz}^0 \approx E_{zx}^0$. It is consistent with the first-principles results that the total ferromagnetic moment vanishes at $\lambda = 0$. By introducing the approximation $E_{yz}^0 = E_{zx}^0 = E_{xy}^0 \equiv E_{t_{2g}}^0$, the orbital magnetic moment can be simplified as

$$
\langle L \rangle_{\text{total}}^{Pn'a'2_1} = -4\sqrt{3} \left(\frac{\zeta}{E_+^0 - E_{t_{2g}}^0} \right) \sin \theta_t \sin \theta_{\text{JT}}.\tag{5}
$$

Interestingly, for the d^4 (Cr²⁺) with $Pna'2'_1$ symmetry, the same total orbital magnetic moment formula is obtained, $\langle L \rangle_{total}^{Pna'2'_{1}}$ = $\langle L \rangle_{\text{total}}^{Pn'a'2_1}$, except that the direction is along the a-axis. Therefore, the same arguments are also valid leading to the same simplified form of Eq. (5). For the d^9 configuration (Cu²⁺), the sign of the orbital magnetic moment is inverted in both magnetic groups.

As a preliminary step for the comparison between the DFT results and the predictions from the model, we parametrize the JT phase

 $\sin \theta_t \sin \theta_{\text{JT}}$ netic moment of reference ions of (a) **1-Cu** and Figure 6: The direction of the local orbital mag-(b) **1-Cr** in their local spherical coordinates (θ) : polar/ ϕ : azimuthal) obtained from DFT and the model. The total orbital magnetic moment of (c) 1 -Cu and (d) 1 -Cr obtained from DFT and model. Model values in (c) and (d) are fitted to the DFT results (gray dotted line) and evaluated from reasonable physical parameters (black solid line).

of a reference Cr/Cu ion as a function of λ , $\tan(\pi - \theta_{\text{JT}})/\tan(\pi - \theta_{\text{JT},\lambda=1}) = \lambda$, where $\theta_{\text{JT},\lambda=1}$ is the JT phase at $\lambda=1$ (See Sec. S9) for the details). For simplicity, let us consider $\theta_{\text{JT},\lambda=1} = 2\pi/3$. The JT phase becomes $\theta_{\text{JT}} = \pi - \tan^{-1}(\sqrt{3}\lambda)$. As a result, the simplified orbital magnetic moment Eq. (5) can be written in terms of the λ ,

$$
\langle L \rangle_{\text{total}}^{d^4/d^9} = \mp 4\sqrt{3} \left(\frac{\zeta}{E_+^0 - E_{t_{2g}}^0} \right) \sin \theta_t \frac{\sqrt{3}\lambda}{\sqrt{3\lambda^2 + 1}}. \tag{6}
$$

Ignoring the JT phase dependency of E_i^0 's, we define the λ independent factor of this expression as $A = \pm 4\sqrt{3}(\zeta/(E_{+}^0 - E_{t_{2g}}^0))\sin\theta_t$. To check the validity of our approach, we compare the model with the DFT values by fitting the single parameter A. Fitted A values are 0.090 for 1-Cu and -0.033 for 1-Cr. In addition, the A values are estimated from physical parameters as 0.082 for **1-Cu** and -0.040 for **1-Cr**,

Figure 7: The spin textures of (a) **1-Cu** and (b) **1-Cr**, at the $k_z = 0$ (upper panels) and $k_z = \pi/c$ (lower panels) planes. The in-plane x and y components are represented by arrows of which the length is the in-plane magnitude with respect to the reference spin $1/2$ above the figure. The z components are represented by a colormap of dots. The inner boundary of each figure coincides with the Brillouin zone boundary.

which are in good agreement with the fitted values (See Sec. S9). As shown in Fig. 6 (c) and (d), the model with these A values well explains the DFT results.

As a further step, let us consider the direction of the local orbital magnetic moment of a reference Cu and Cr ions expected from Eq. (3), as shown in Fig. 6 (a) and (b) in the spherical coordinates with respect to their local orbital-axes, respectively. The directions from the DFT are shown together for comparison. Except for the deviation in the polar angle θ of **1-Cu**, the model well predicts the orbital magnetic moment direction. The deviation may originate from oversimplification of the model, such as ignoring the θ_{JT} -dependence of orbital energies, degeneracy, and higher orders in ζ.

Interestingly, the model accounts well for the finite orbital magnetic moment, although we as-

sumed the uncanted spin configuration in which the total spin moment vanishes. This is possible because the spin and orbital moments are not parallel in the $1-Cr/Cu$. The orbital moment rotates depending on the JT phase in spite of the fixed spin direction. The inclusion of the small spin canting would not significantly alter the predicted orbital moment ($\sim 1\%$). In other words, the orbital moment is insensitive to a moderate spin fluctuation, i.e., the orbital moment is robust. Moreover, the orbital magnetic moment is linear to the SOC strength ζ , whereas the MSIA is second order (Sec. S10). These properties provide a robust justification of WFM in terms of the total magnetic moment $(\text{spin} + \text{orbital})$ for **1-Cu** and also emphasize the role of the JT effect on it. Our prediction might be confirmed by an x-ray magnetic circular dichroism experiment, ⁶⁰ which can measure the orbital magnetic moment.

Finally, we remark that the complex interplay between the SOC and JT distortion may induce unusual spin and orbital textures in the k-space. The spin texture $(\mathbf{s}_{\mathbf{k}n} = \langle \psi_{\mathbf{k}n} | \mathbf{S} | \psi_{\mathbf{k}n} \rangle$, where n is a band index) of the highest occupied band of each compound at $k_z = 0$ and $k_z = \pi/c$ planes are shown in Fig. 7. In the 1 -Cu, a 'curly' in-plane texture appears around the center at $k_z = 0$ plane. On the other hand, an irregular texture mostly aligned to z-direction can be seen at $k_z = \pi/c$ plane. In the **1-Cr**, persistent type spin textures⁶¹ appear for both the $k_z = 0$ and $k_z = \pi/c$ planes. Since the orbital magnetic moment is non-negligible in the 1-Cu, it is worth investigating the orbital texture in the $1-Cr/Cu$. For example, we found Dresselhaustype⁶² orbital texture in the **1-Cr** (See Sec. S11) in SI).

Conclusion

In this work, we reported for the first time the synthesis of $1-Cr$, a missing member of $[C(NH₂)₃]M(HCOO)₃$ family. We identified that the 1-Cr has perovskite type JT distorted polar $Pna2₁$ structure as initially predicted, as well as the AFM ordering below 8 K. We provided an improved understanding of the magnetic properties of $1-Cr/Cu$ by the firstprinciples calculation and model studies which provide a better estimate of the T_c by including the on-site Coulomb energy corrections in DFT. The revised magnetic moment estimation is also more consistent with the experiment. In addition, new calculations suggested potential importance of the orbital magnetic moment in the $1-Cr/Cu$, i.e., the contribution of the orbital magnetic moment to the net ferromagnetic moment is even larger than the spin contribution in the 1-Cu. We revealed that the orbital magnetic moment is robust and explicitly coupled with the JT distortion by the model study. This approach can be generally considered in the d^4/d^9 JT-systems.

Associated Contents

Supporting Information

The Supporting Information is available free of charge at [URL will be added by publisher].

Additional Sample Preparation Details; Single Crystal Diffraction Measurements and Analysis; Symmetry Analysis of Magnetic Ground State; Spin Model; Monte Carlo Results; DFT Parameter Dependence of Magnetic Moments; Matrix Representations of the Spin-orbit Coupling; Calculation of The Perturbed Orbital Angular Momentum; Details in The Model Analysis for $1-Cr/Cu$; The Second Order Energy Correction Term in The Perturbation Theory; Spin and Orbital Textures in k-space (PDF)

Author Information

Corresponding Authors

Jeajun Yu - jyu@snu.ac.kr Paul J. Saines - P.Saines@kent.ac.uk Alessandro Stroppa - alessandro.stroppa@spin.cnr.it

Notes

The authors declare no competing financial interest.

Acknowledgement

A.S. thanks Prof. Holger Theisel for useful discussions on graphical rendering of twodimensional vector fields, i.e. spin-textures. J.Y. acknowledges financial support from Samsung Electronics Co., Ltd. K.Y. was supported in part by a KIAS individual Grant (No. CG092501). P.J.S. acknowledges the support from Engineering and Physical Science Research Council, United Kingdom Research and Innovation via project EP/R011524/1.

References

- (1) Furukawa, H.; Cordova, K. E.; O'Keeffe, M.; Yaghi, O. M. The Chemistry and Applications of Metal-Organic Frameworks. Science **2013**, 341, 1230444.
- (2) Suh, M. P.; Park, H. J.; Prasad, T. K.; Lim, D.-W. Hydrogen Storage in Metal–Organic Frameworks. Chem. Rev. 2012, 112, 782–835.
- (3) Cheetham, A. K.; Rao, C. N. R. There's Room in the Middle. Science 2007, 318, 58–59.
- (4) Wang, X.-Y.; Gan, L.; Zhang, S.-W.; Gao, S. Perovskite-like Metal Formates with Weak Ferromagnetism and as Precursors to Amorphous Materials. Inorg. *Chem.* **2004**, $\frac{43}{4}$, 4615–4625.
- (5) Ye, Q.; Song, Y.-M.; Wang, G.-X.; Chen, K.; Fu, D.-W.; Hong Chan, P. W.; Zhu, J.-S.; Huang, S. D.; Xiong, R.-G. Ferroelectric Metal-Organic Framework with a High Dielectric Constant. J. Am. Chem. Soc. 2006, 128, 6554–6555.
- (6) Van Aken, B. B.; Palstra, T. T.; Filippetti, A.; Spaldin, N. A. The origin of ferroelectricity in magnetoelectric $YMnO₃$. Nature Mater 2004, 3, 164–170.
- (7) Cheong, S.-W.; Mostovoy, M. Multiferroics: a magnetic twist for ferroelectricity. Nature Mater 2007, 6, 13–20.
- (8) Malashevich, A.; Vanderbilt, D. First Principles Study of Improper Ferroelectricity in Tb MnO_3 . Phys. Rev. Lett. 2008, 101, 037210.
- (9) Stroppa, A.; Jain, P.; Barone, P.; Marsman, M.; Perez-Mato, J. M.; Cheetham, A. K.; Kroto, H. W.; Picozzi, S. Electric Control of Magnetization and Interplay between Orbital Ordering and Ferroelectricity in a Multiferroic Metal-Organic Framework. Angew. Chem. Int. Ed. 2011, 50, 5847-5850.
- (10) Picozzi, S.; Stroppa, A. Advances in abinitio theory of multiferroics: Materials and mechanisms: modelling and understanding. Eur. Phys. J. B 2012, 85, 240.
- (11) Zhang, W.; Xiong, R.-G. Ferroelectric Metal-Organic Frameworks. Chem. Rev. 2012, 112, 1163–1195.
- (12) Stroppa, A.; Barone, P.; Jain, P.; Perez-Mato, J. M.; Picozzi, S. Hybrid Improper Ferroelectricity in a Multiferroic and Magnetoelectric Metal-Organic Framework. Adv. Mater. 2013, 25, 2284–2290.
- (13) Di Sante, D.; Stroppa, A.; Jain, P.; Picozzi, S. Tuning the Ferroelectric Polarization in a Multiferroic Metal–Organic Framework. J. Am. Chem. Soc. 2013, 135, 18126–18130.
- (14) Tian, Y.; Stroppa, A.; Chai, Y.-S.; Barone, P.; Perez-Mato, M.; Picozzi, S.; Sun, Y. High-Temperature Ferroelectricity and Strong Magnetoelectric Effects in a Hybrid Organic-Inorganic Perovskite Framework: High-Temperature Ferroelectricity and Strong Magnetoelectric Effects in a Hybrid Organic-Inorganic Perovskite Framework. Phys. Status Solidi RRL 2015, 9, 62–67.
- (15) Gómez-Aguirre, L. C.; Pato-Doldán, B.; Stroppa, A.; Yang, L.-M.; Frauenheim, T.; Mira, J.; Yáñez Vilar, S.; Artiaga, R.; Castro-García, S.; Sánchez-Andújar, M.; Señarís Rodríguez, M. A. Coexistence of Three Ferroic Orders in the Multiferroic Compound $[(CH_3)_4N][Mn(N_3)_3]$ with Perovskite-Like Structure. Chem. Eur. J. 2016, 22, 7863–7870.
- (16) Evans, N. L.; Thygesen, P. M. M.; Boström, H. L. B.; Reynolds, E. M.; Collings, I. E.; Phillips, A. E.; Goodwin, A. L. Control of Multipolar and Orbital Order in Perovskitelike $[C(NH_2)_3]Cu_xCd_{1-x}(HCOO)_3$ Metal–Organic Frameworks. 138, 9393– 9396.
- (17) Fan, F.-R.; Wu, H.; Nabok, D.; Hu, S.; Ren, W.; Draxl, C.; Stroppa, A. Electric-Magneto-Optical Kerr Effect in a Hybrid Organic–Inorganic Perovskite. J. Am. Chem. Soc. 2017, 139, 12883–12886.
- (18) Saines, P. J.; Bristowe, N. C. Probing magnetic interactions in metal–organic frameworks and coordination polymers microscopically. Dalton Trans. 2018, 47, 13257–13280.
- (19) Ma, Y.; Sun, Y. Multiferroic and thermal expansion properties of metal-organic frameworks. J. A ppl. Phys. 2020, 127, 080901.
- (20) Kanižaj, L.; Barišić, D.; Torić, F.; Pajić, D.; Molčanov, K.; Šantić, A.; Lončarić, I.; Jurić, M. Structural, Electrical, and Magnetic Versatility of the Oxalate-Based [CuFe] Compounds Containing 2,2':6',2"-Terpyridine: Anion-Directed Synthesis. Inorg. Chem. 2020, 59, 18078–18089.
- (21) Šenjug, P.; Dragović, J.; Kalanj, M.; Torić, F.; Rubčić, M.; Pajić, D. Magnetic behaviour of $(C_2H_5NH_3)_2CuCl_4$ type multiferroic. J. Magn. Magn. Mater. 2019, 479, 144–148.
- (22) Yang, Z.; Cai, G.; Bull, C. L.; Tucker, M. G.; Dove, M. T.; Friedrich, A.; Phillips, A. E. Hydrogen-bond-mediated structural variation of metal guanidinium formate hybrid perovskites under pressure. 377, 20180227.
- (23) Gonçalves, J. N.; Phillips, A. E.; Li, W.; Stroppa, A. First-Principles Study of Structure and Magnetism in Copper(II)- Containing Hybrid Perovskites. 10, 1129.
- (24) Kang, S.; Yu, J. Electronic structure and magnetic properties of transition metal kagome metal–organic frameworks. Phys. Chem. Chem. Phys. 2022, 24, 22168– 22180.
- (25) Hu, K.-L.; Kurmoo, M.; Wang, Z.; Gao, S. Metal-Organic Perovskites: Synthesis, Structures, and Magnetic Properties of $\left[\text{C(NH }_{2})_{3}\right][\text{M}^{\text{II}}(\text{HCOO})_{3}]$ (M=Mn, Fe, Co, Ni, Cu, and Zn; $C(NH_2)_3 = \text{Guani}$ dinium). Chem. Eur. J. 2009, 15, 12050– 12064.
- (26) Benedek, N. A.; Fennie, C. J. Hybrid Improper Ferroelectricity: A Mechanism for Controllable Polarization-Magnetization Coupling. Phys. Rev. Lett. 2011, 106, 107204.
- (27) Rondinelli, J. M.; Fennie, C. J. Octahedral Rotation-Induced Ferroelectricity in Cation Ordered Perovskites. Adv. Mater. 2012, 24, 1961–1968.
- (28) Benedek, N. A.; Rondinelli, J. M.; Djani, H.; Ghosez, P.; Lightfoot, P. Understanding ferroelectricity in layered perovskites: new ideas and insights from theory and experiments. Dalton Trans. 2015, 44, 10543–10558.
- (29) Boström, H. L. B.; Senn, M. S.; Goodwin, A. L. Recipes for improper ferroelectricity in molecular perovskites. Nat Commun 2018, 9, 2380.
- (30) Ai, Y.; Sun, R.; Liao, W.-Q.; Song, X.- J.; Tang, Y.-Y.; Wang, B.-W.; Wang, Z.- M.; Gao, S.; Xiong, R.-G. Unprecedented Ferroelectricity and Ferromagnetism in a Cr2+ Based Two-Dimensional Hybrid Perovskite. Angew. Chem. Int. Ed. 2022, 61 .
- (31) Ai, Y.; Sun, R.; Zeng, Y.-L.; Liu, J.-C.; Tang, Y.-Y.; Wang, B.-W.; Wang, Z.-M.; Gao, S.; Xiong, R.-G. Coexistence of magnetic and electric orderings in a divalent Cr^{2+} -based multiaxial molecular ferroelectric. Chem. Sci. 2021, 12, 9742–9747.
- (32) Brauer, G., Ed. Handbook of Preparative Inorganic Chemistry (Second Edition); Academic Press, 1963.
- (33) Howard, C. J.; Hunter, B. A. A Computer Program for Rietveld Analysis of X-ray

and Neutron Powder Diffraction Patterns. 1998.

- (34) Kresse, G.; Hafner, J. Ab initio molecular dynamics for liquid metals. Phys. Rev. B 1993, 47, 558–561.
- (35) Kresse, G.; Hafner, J. Ab initio moleculardynamics simulation of the liquid-metal– Amorphous-semiconductor transition in germanium. Phys. Rev. B $1994, 49$, 14251–14269.
- (36) Kresse, G.; Furthmüller, J. Efficiency of ab-initio total energy calculations for metals and semiconductors using a plane-wave basis set. Comput. Mater. Sci. 1996, 6, 15–50.
- (37) Kresse, G.; Furthmüller, J. Efficient iterative schemes for *ab initio* total-energy calculations using a plane-wave basis set. Phys. Rev. B **1996**, 54, 11169–11186.
- (38) Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized Gradient Approximation Made Simple. Phys. Rev. Lett. 1996, 77, 3865–3868.
- (39) Kresse, G.; Joubert, D. From ultrasoft pseudopotentials to the projector augmented-wave method. Phys. Rev. B 1999, 59, 1758–1775.
- (40) Capillas, C.; Tasci, E. S.; Flor, G. d. l.; Orobengoa, D.; Perez-Mato, J. M.; Aroyo, M. I. A new computer tool at the Bilbao Crystallographic Server to detect and characterize pseudosymmetry. Z. Kristallogr. Cryst. Mater. 2011, 226, 186– 196.
- (41) Liechtenstein, A. I.; Anisimov, V. I.; Zaanen, J. Density-functional theory and strong interactions: Orbital ordering in Mott-Hubbard insulators. Phys. Rev. B 1995, 52, R5467–R5470.
- (42) Brese, N. E.; O'Keeffe, M. Bond-valence parameters for solids. Acta Cryst B 1991, 47, 192–197.
- (43) Fackler, J. P. J.; Holah, D. G. Properties of Chromium(II) Complexes. I. Electronic Spectra of the Simple Salt Hydrates. Inorganic Chemistry 1965, 4, 954–958.
- (44) Aakesson, R.; Pettersson, L. G. M.; Sandstroem, M.; Wahlgren, U. Theoretical calculations of the Jahn-Teller effect in the hexahydrated Copper(II), chromium(II), and manganese(III) ions, hexaaquacop $per(2+)$, hexaaquachromium $(2+)$ and hexaaquamanganese $(3+)$, and comparisons with the hexahydrated copper (I) , chromium(III), and manganese(II) clusters. The Journal of Physical Chemistry 1992, 96, 150–156.
- (45) Collings, I. E.; Hill, J. A.; Cairns, A. B.; Cooper, R. I.; Thompson, A. L.; Parker, J. E.; Tang, C. C.; Goodwin, A. L. Compositional dependence of anomalous thermal expansion in perovskite-like ABX_3 formates. Dalton Trans. 2016, 45, 4169–4178.
- (46) Burley, L. G.; Beecham-Lonsdale, J. H.; Srivastava, A. K.; Collings, I. E.; Saines, P. J. Enhancing the chemical flexibility of hybrid perovskites by introducing divalent ligands. Dalton Trans. 2021, 50, 5437–5441.
- (47) Vasiliev, A. N.; Volkova, O. S.; Hammer, E.; Glaum, R.; Broto, J.-M.; Millot, M.; Nénert, G.; Liu, Y. T.; Lin, J.- Y.; Klingeler, R.; Abdel-Hafiez, M.; Krupskaya, Y.; Wolter, A. U. B.; Büchner, B. Weak ferrimagnetism and multiple magnetization reversal in α -Cr₃(PO₄)₂. Phys. Rev. B 2012, 85, 014415.
- (48) Tezuka, K.; Shan, Y. J.; Imoto, H.; Ohoyama, K. Crystal and magnetic structures of Y_2CrS_4 . J. Phys. Chem. Solids 2007, 68, 2133–2137.
- (49) Goodenough, J. B. Magnetism and the chemical bond; R. E. Krieger Pub. Co, 1976.
- (50) Khomskii, D. I. Transition Metal Compounds; Cambridge University Press, 2014.
- (51) Šenjug, P.; Dragović, J.; Torić, F.; Lončarićć, I.; Despoja, V.; Smokrović, K.; Topić, E.; Dilović, I.; Rubčić, M.; Pajić, D. Magnetoelectric Multiferroicity and Magnetic Anisotropy in Guanidinium Copper(II) Formate Crystal. Materials 2021, 14, 1730.
- (52) Moriya, T. Theory of Magnetism of Nif_2 . Phys. Rev. 1960, 117, 635–647.
- (53) Bruno, P. Tight-binding approach to the orbital magnetic moment and magnetocrystalline anisotropy of transitionmetal monolayers. Phys. Rev. B 1989, 39, 865–868.
- (54) Blanco-Rey, M.; Cerdá, J. I.; Arnau, A. Validity of perturbative methods to treat the spin–orbit interaction: application to magnetocrystalline anisotropy. New J. Phys. 2019, 21, 073054.
- (55) Takayama, H.; Bohnen, K.-P.; Fulde, P. Magnetic surface anisotropy of transition metals. Phys. Rev. B 1976, 14, 2287– 2295.
- (56) Jahn, H.; Teller, E. Stability of polyatomic molecules in degenerate electronic states-I Orbital degeneracy. Proc. R. Soc. Lond. A 1937, 161, 220–235.
- (57) Bersuker, I. B. Modern Aspects of the Jahn-Teller Effect Theory and Applications To Molecular Problems. Chem. Rev. 2001, 101, 1067–1114.
- (58) Pavarini, E.; Koch, E.; Anders, F.; Jarrell, M. Correlated Electrons: from Models to Materials: Lecture Notes of the Autumn School Correlated Electrons 2012 at Forschungszentrum Jülich, 3-7 September 2012 ; Forschungszentrum Jülich, Zentralbibliothek, Verlag, 2012; Chapter 6.
- (59) Stroppa, A.; Barone, P.; Di Sante, D.; Cuoco, M.; Picozzi, S.; Whangbo, M.- H. Analogies between Jahn-Teller and Rashba spin physics. Int. J. Quantum Chem. 2016, 116, 1442–1450.
- (60) Thole, B. T.; Carra, P.; Sette, F.; van der Laan, G. X-ray circular dichroism as a probe of orbital magnetization. Phys. Rev. Lett. **1992**, 68, 1943–1946.
- (61) Tao, L. L.; Tsymbal, E. Y. Persistent spin texture enforced by symmetry. Nat. Commun. **2018**, 9, 2763.
- (62) Winkler, R. In Spin-Orbit Coupling Effects in Two-Dimensional Electron and Hole Systems; Winkler, R., Ed.; Springer Tracts in Modern Physics; Springer, pp 69–129.

TOC Graphic

For Table of Contents Only

Synopsis

Among the multiferroic metal-organic framework $[C(NH₂)₃]M(HCOO)₃$ family, where M is a 3d-transition metal, the $M = Cr$ case was missing for more than 10 years, although its existence and properties are predicted in 2013. We report the synthesis, structure, and magnetic properties of the $M = Cr$ case, $[C(NH₂)₃]Cr(HCOO)₃$. Density functional theory calculations and model studies support our observation. In addition, the $M = Cu$ case is suggested to exhibit orbital-dominant weak ferromagnetism.